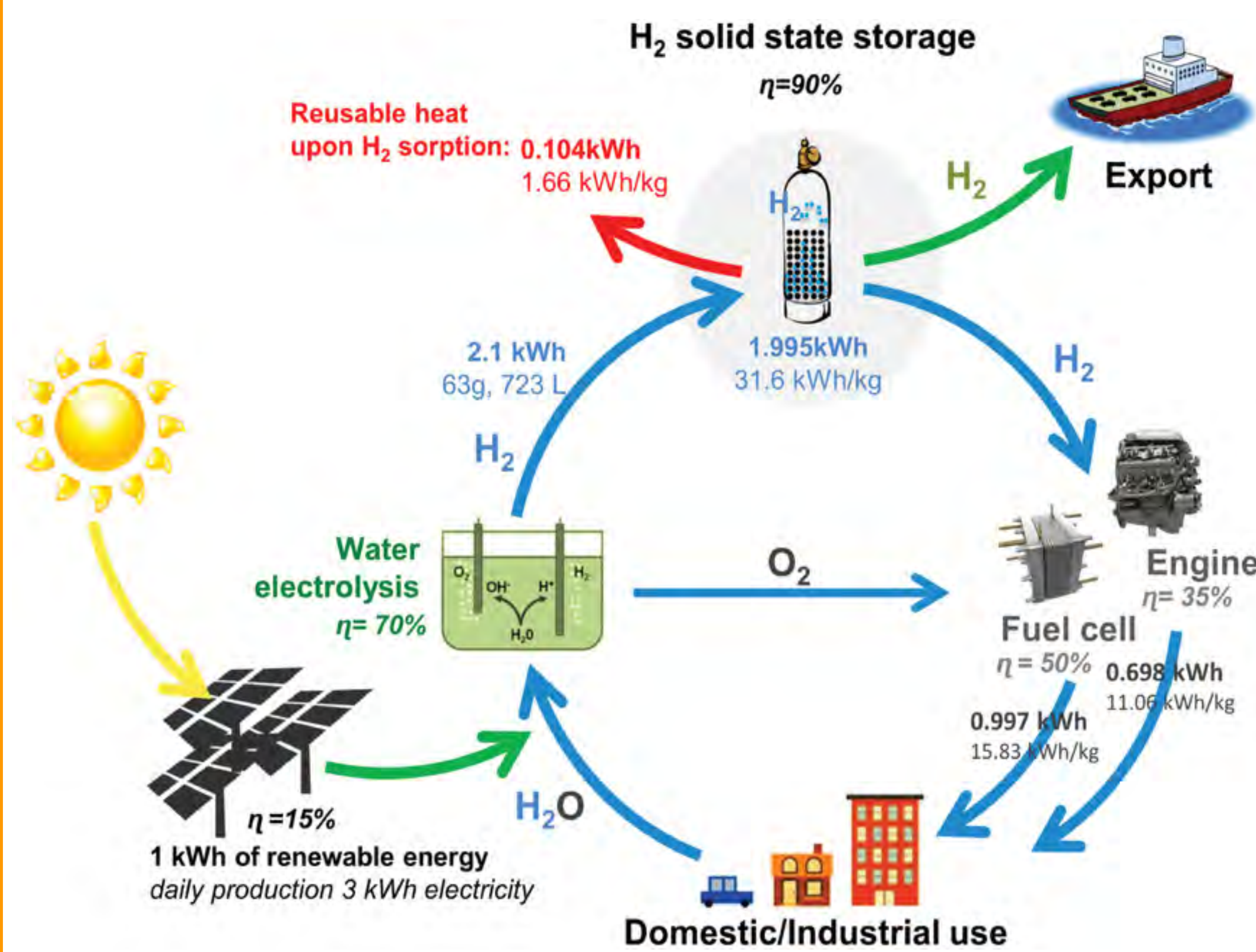


Boosting H₂ Release in NaBH₄ through Structural Control

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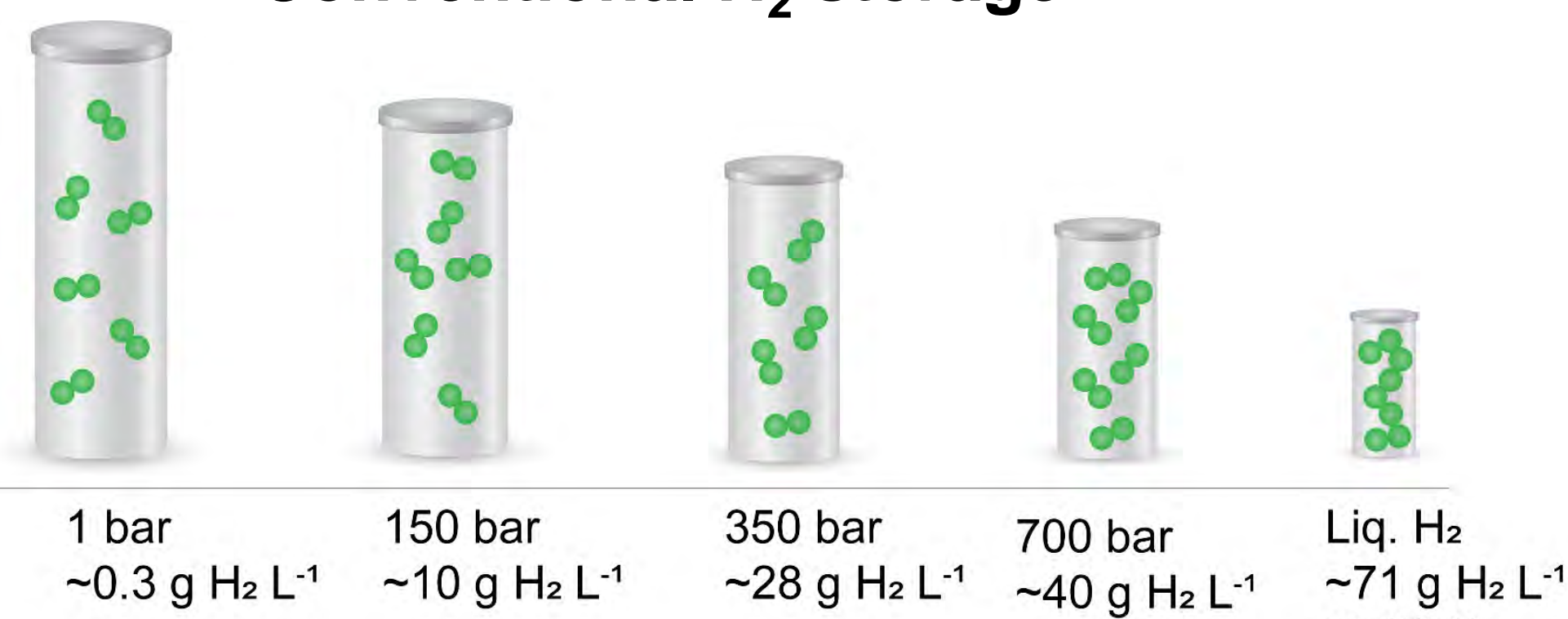
Hydrogen – A Key Contributor to the Energy Transition



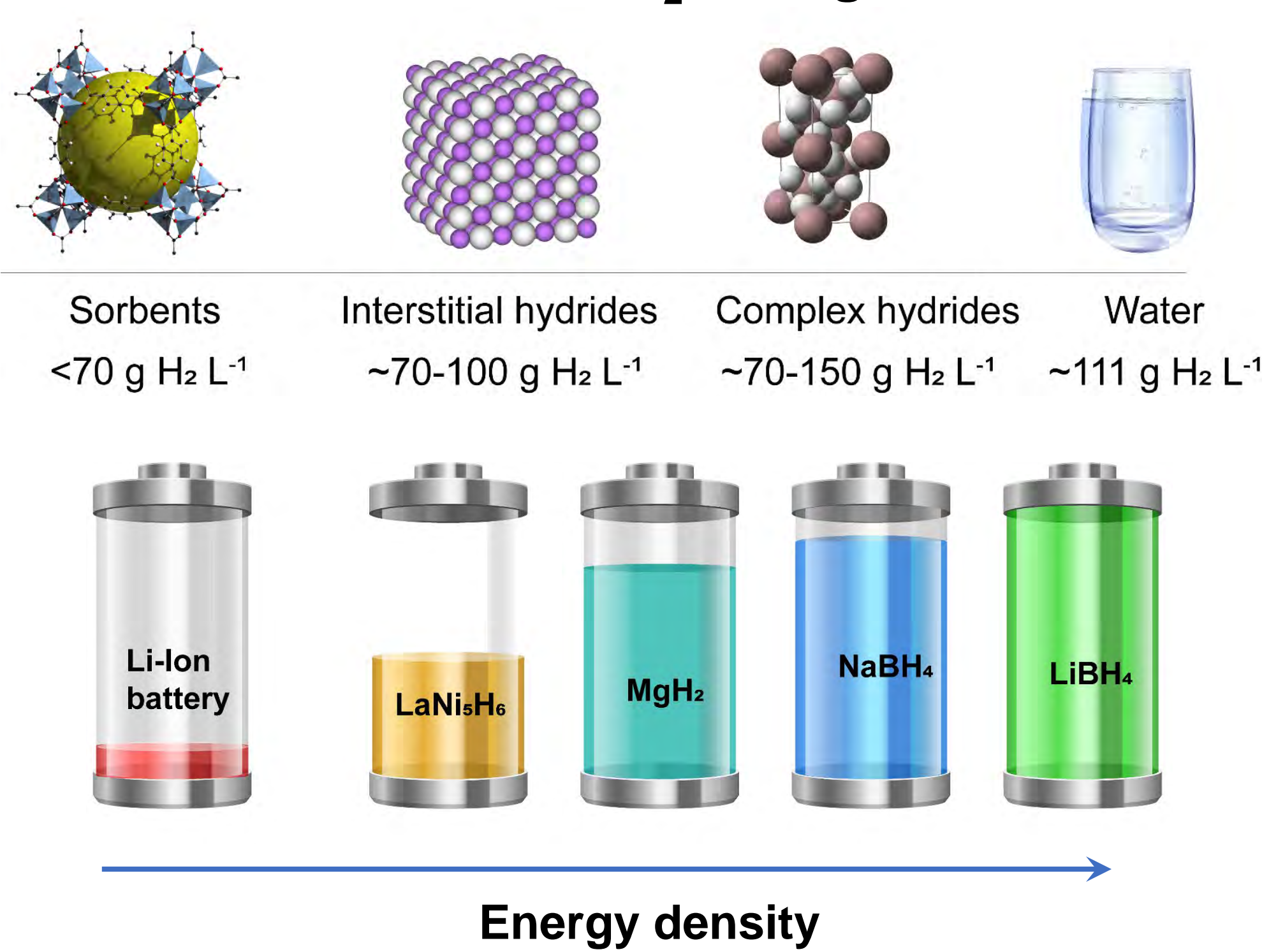
- Hydrogen is a potential energy vector
- Provides more reliability and flexibility in enabling the use of renewables across industries
- Hydrogen production, storage and distribution/utilization are the key players for the global hydrogen economy
- The most demanding – H₂ storage

Hydrogen Storage – Quo Vadis?

Conventional H₂ storage

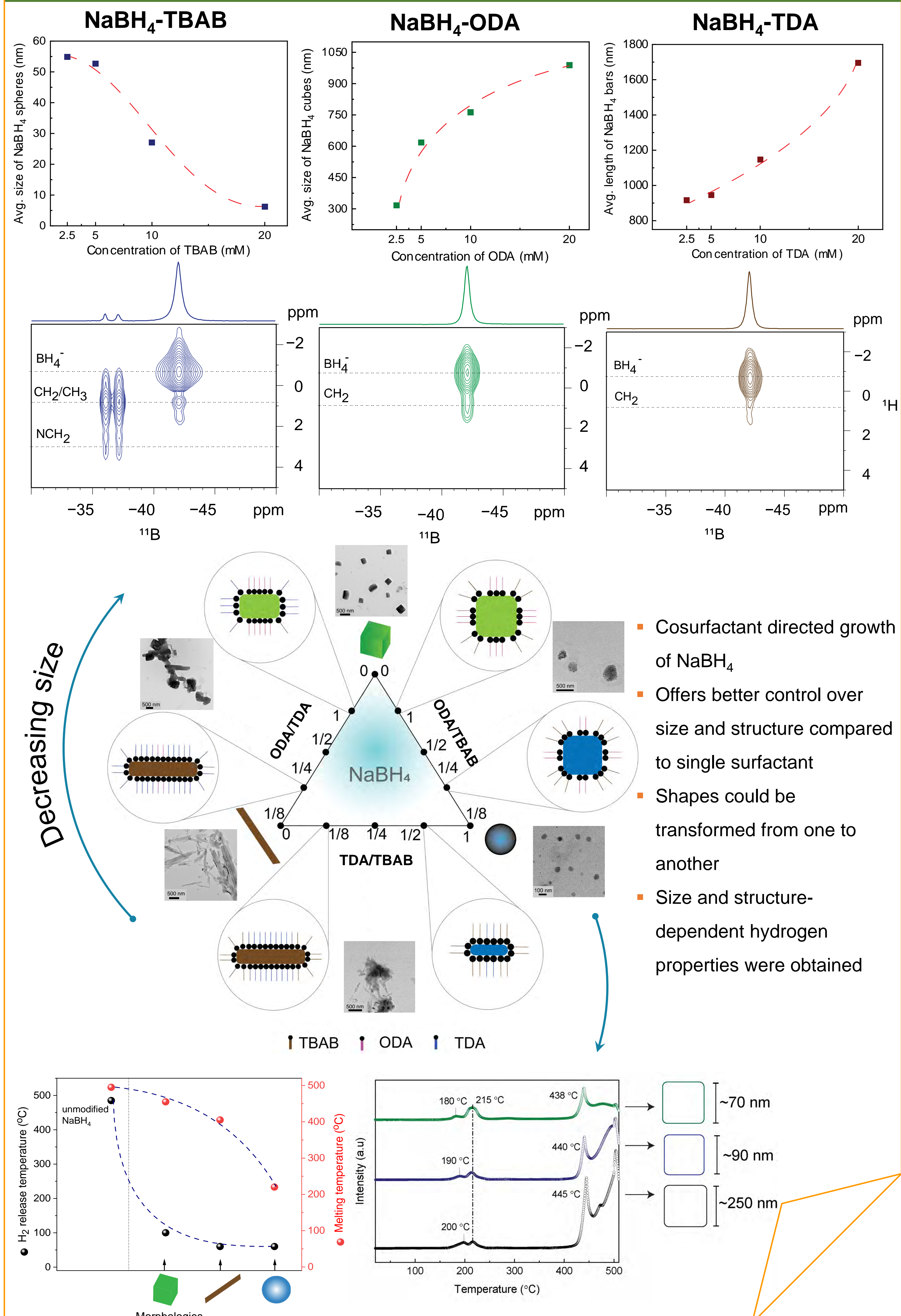


Material-based H₂ storage

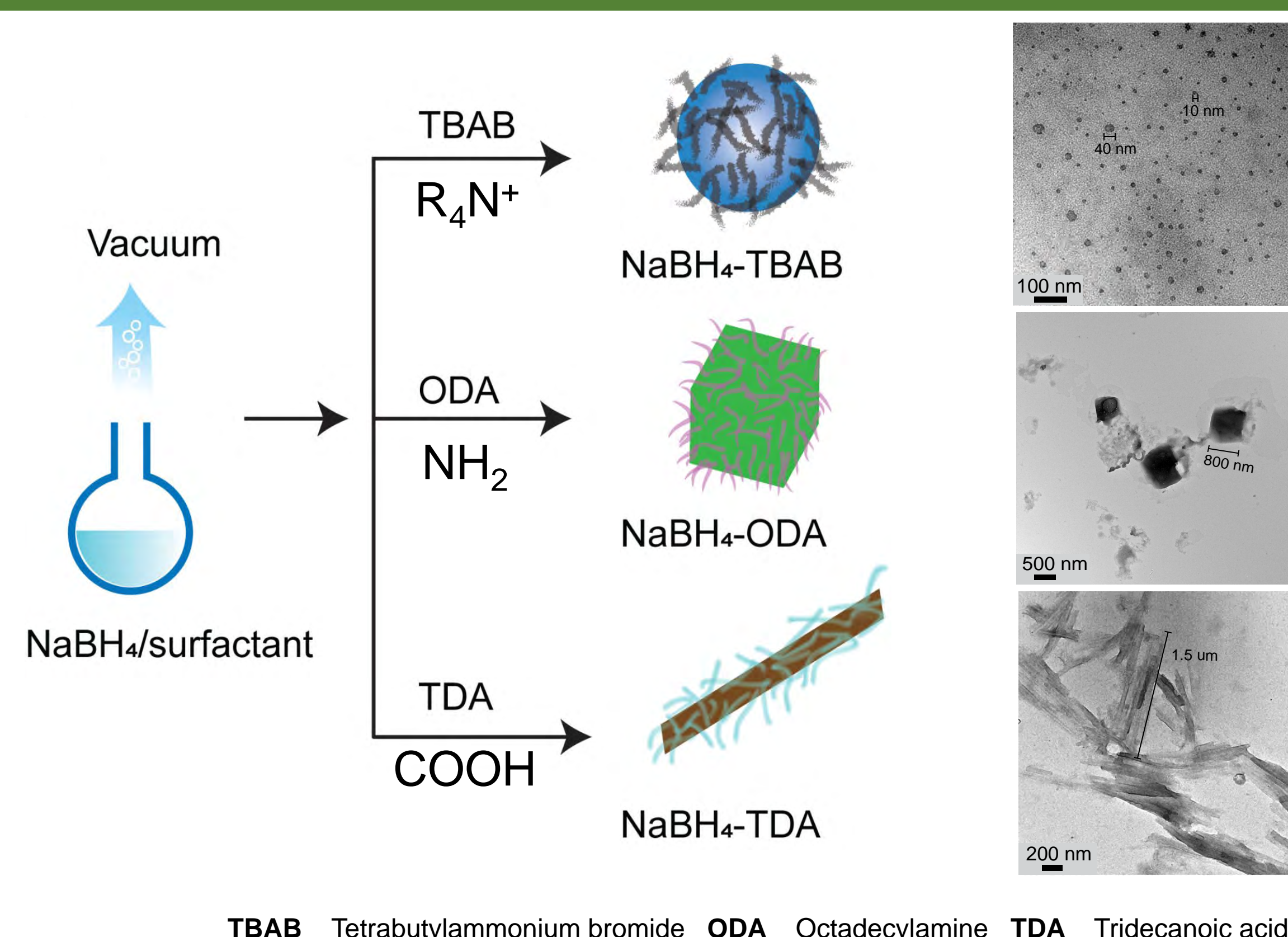


- Gaseous-state H₂ storage; the most mature and simple technique to store hydrogen
- Material-based H₂ storage is advantageous – offers higher energy densities and safety
- Borohydrides are potential candidates. NaBH₄ with ~10.6 mass% H₂ is most widely used but requires >500 °C to release H₂
- Strategies to destabilize NaBH₄ are needed
- Structure-dependent H₂ release has never been realized
- Nanosizing has the potential to overcome the thermodynamic and kinetic barriers

Key Findings



First-step Toward Tailoring NaBH₄ at the Nanoscale



TBAB Tetrabutylammonium bromide ODA Octadecylamine TDA Tridecanoic acid

- Synthesis via evaporation induced self-assembly approach – offers higher flexibility and tunability compared to other methods
- TBAB, ODA, and TDA directed NaBH₄ spheres, cubes, and bars, respectively.

Conclusion

- For the first time, size and shape-controlled growth of NaBH₄ has been achieved
- The size can be controlled by tuning the synthetic parameters; evaporation rate or concentration of surfactant
- The desired assemblies either as spheres, cubes, or bars were further tailored using a cosurfactant
- The structurally controlled NaBH₄ showed notable shifts in their melting points and H₂ release temperatures
- Further structure-property correlations are anticipated in the future

References

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