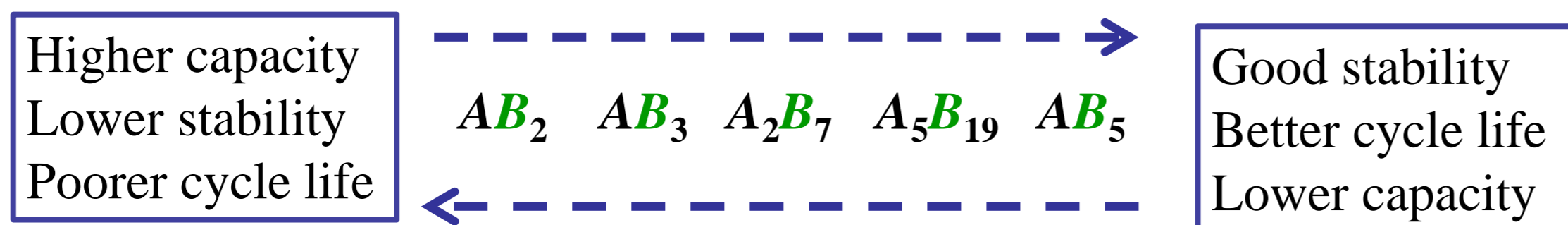


INTRODUCTION

Hydride forming AB_2 -type alloys with C15 Laves phase structure ($Fd\bar{3}m$) possess more tetrahedral sites than AB_3 and AB_n ($3 \leq n \leq 3.8$) ones to accommodate atomic hydrogen, thus higher hydrogen absorption capacity is foreseen [1] (Fig. 1).

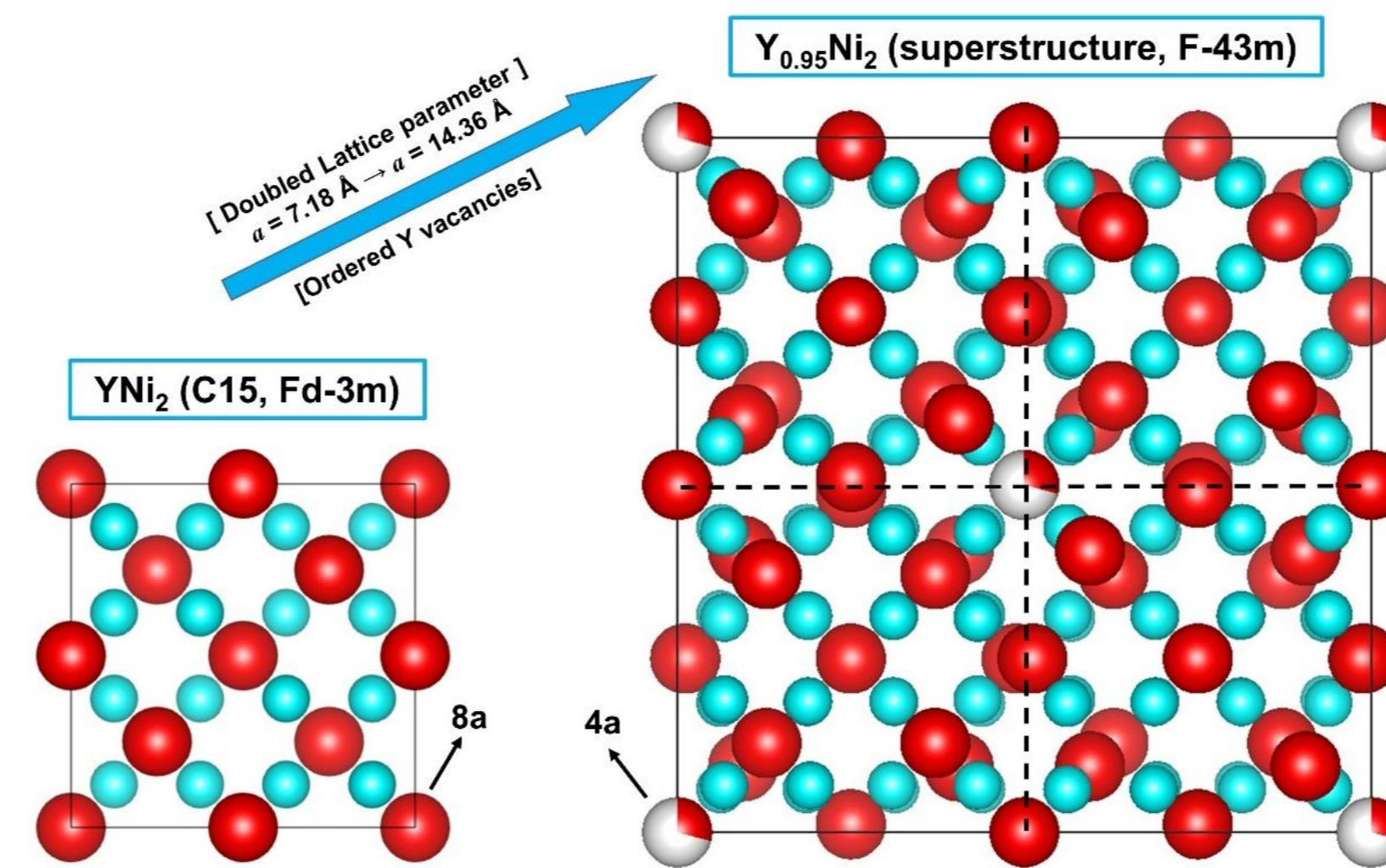
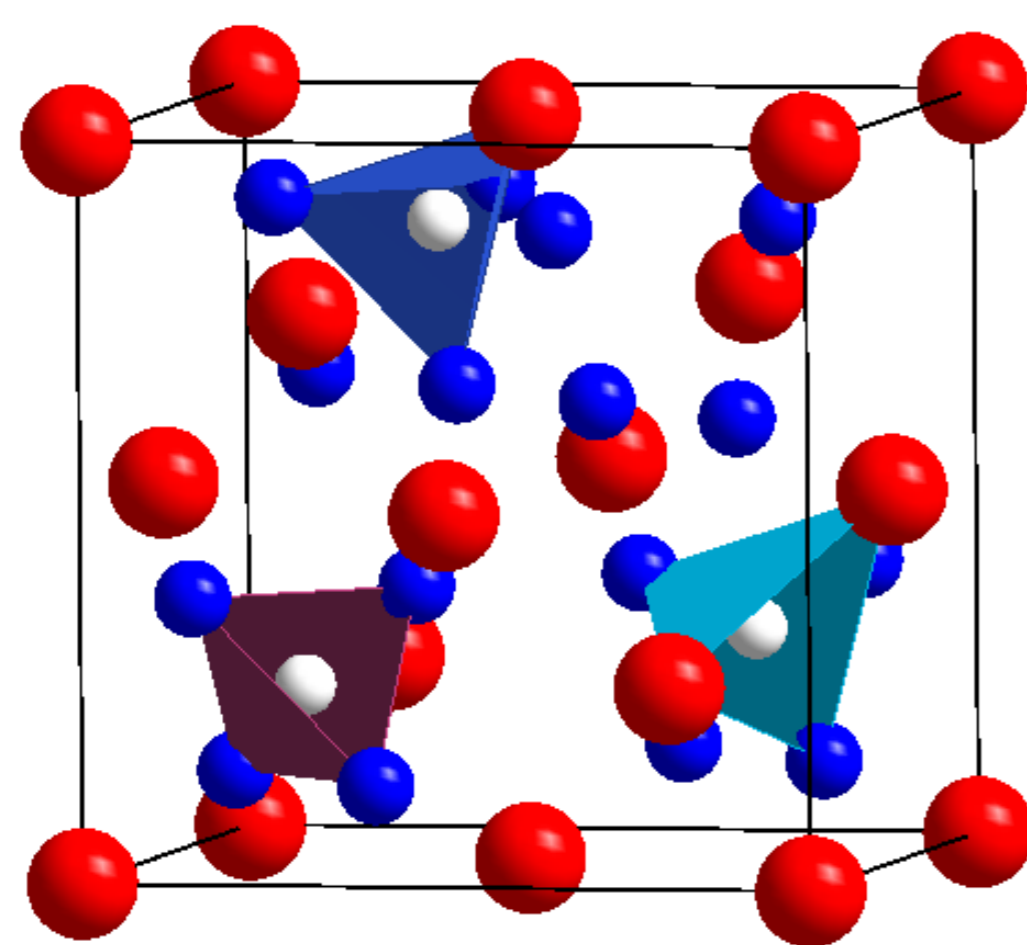


YNi_2 suffers from hydrogen-induced amorphization [2]. This work aims at stabilizing YNi_2 upon hydrogen cycling by substituting Mn for Ni in pseudo-binary $Y_{0.9}Ni_{2-y}Mn_y$ ($0.1 \leq y \leq 0.5$) compounds.

C15 Laves Phase structure showing tetrahedral interstitial sites for H

- Rare earth (R)
- Transition metals (M)
- Hydrogen

The trend of hydrogen occupy at tetrahedral sites:
96g [R2M2] > 32e [RM3] > 8b [M4]



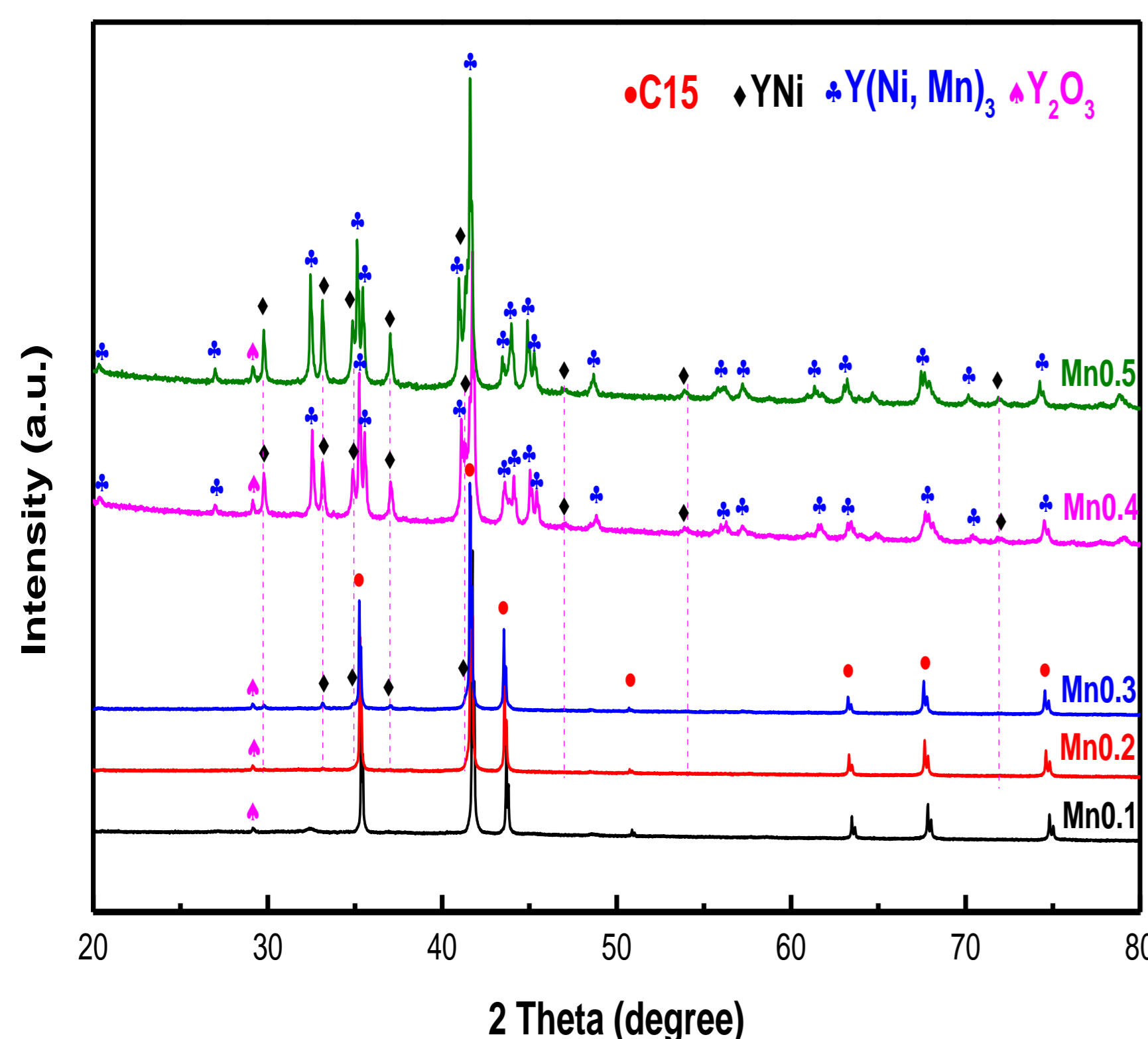
METHODS

- Induction melting: Nominal composition $Y_{0.9}Ni_{2-y}Mn_y$ ($y = 0.1, 0.2, 0.3, 0.4, 0.5$)
- Heat treatment: $T = 1123K$, $t = 3$ days
- X-ray diffraction (XRD)
- Electron Probe Micro-Analysis (EPMA)
- Neutron Powder diffraction (NPD)
- Pressure-Composition Isotherm (PCI): $T = 423$ K, Sieverts' method

Phase transformation from C15 Laves structure to the superstructure with ordered Y-site vacancies at 4a site ($F-43m$)

RESULTS

Phase occurrence

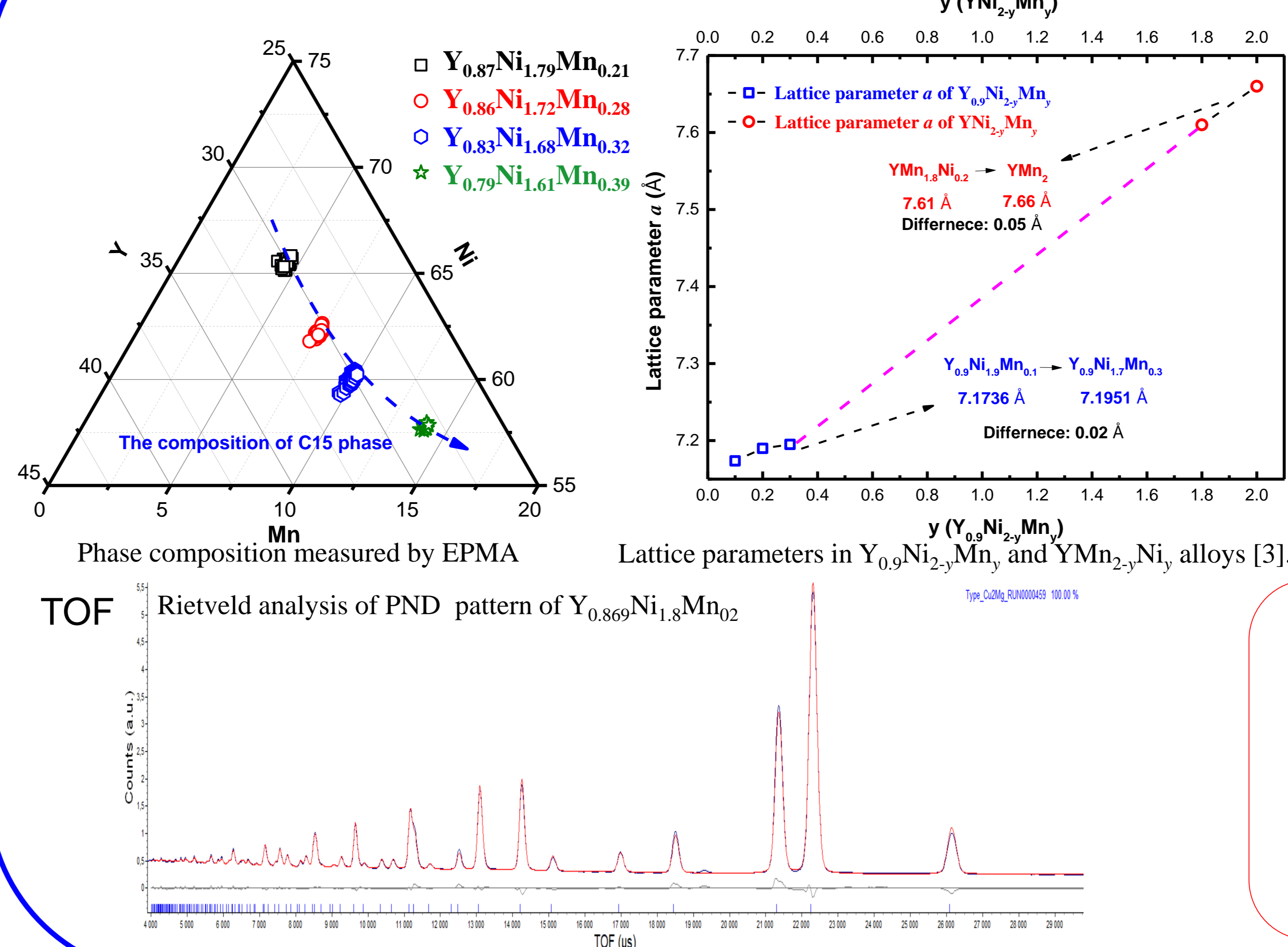


XRD patterns of $Y_{0.9}Ni_{2-y}Mn_y$ ($y = 0.1, 0.2, 0.3, 0.4, 0.5$) alloys

Phase	Mn0.1	Mn0.2	Mn0.3	Mn0.4	Mn0.5
C15	96%	95%	90%	14%	
YNi		4%	9%	23%	27%
AB_3				62%	72%
Y_2O_3	4%	1%	1%	1%	1%

- C15 Laves phase is the main phase for $y \leq 0.3$ and YNi is the secondary phase. The abundance of YNi phase increases with the increasing Mn content to balance the sub-stoichiometry in the C15 phase
- $Y(Ni, Mn)_3$ forms when Mn content up to 0.4 and 0.5, which may indicate the limit of solid solution of Mn in C15 phase.

Structural analysis



- EPMA analyses show Y sub-stoichiometry which increases with Mn content
- However, the lattice parameter of the C15 phase increases gradually with Mn content

$Y_{0.9}Ni_{2-y}Mn_y$ compounds crystallize in the C15 structure with higher content of disordered Y-site vacancies while Mn substitutes Ni?

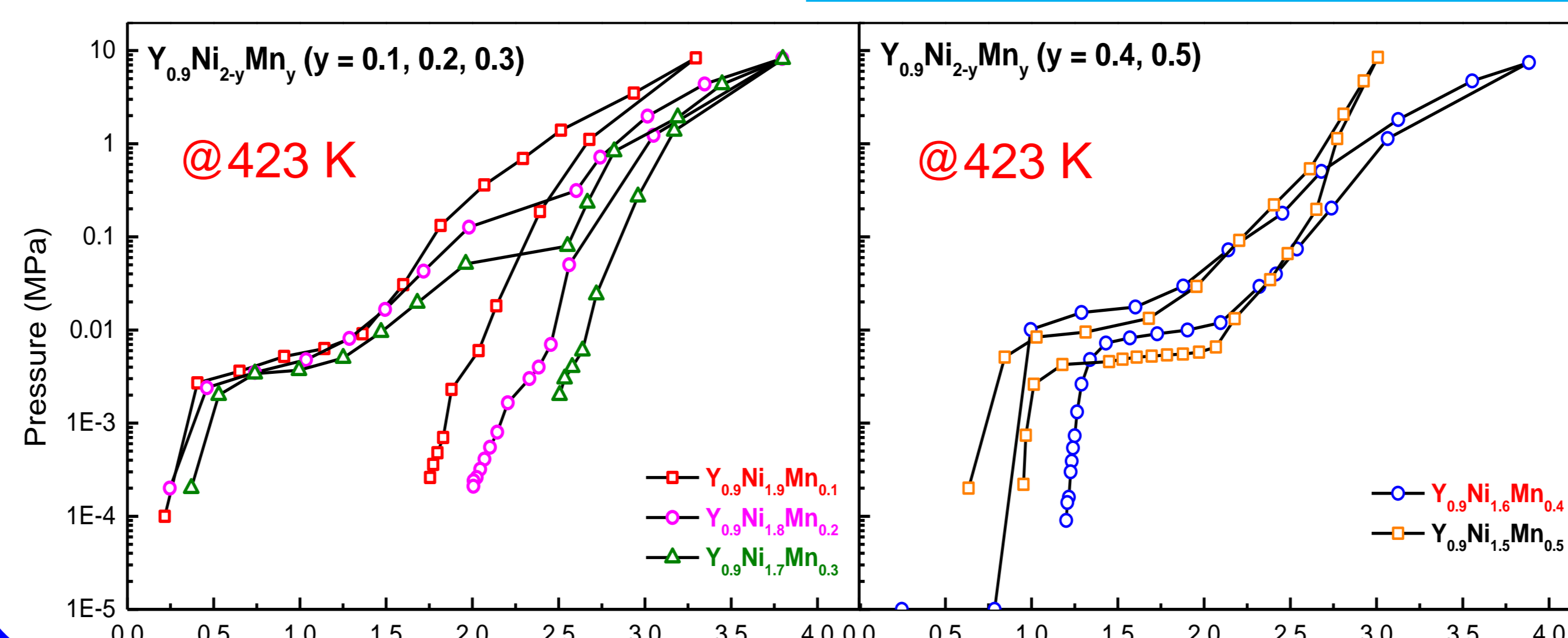
The question cannot be solved by XRD since Ni and Mn cannot be differentiated

Rietveld analysis of PND data shows

- Non vacancies on the A-site
- Mn occupies the B-site as well as the A-site.

Mn behaves both as a A- and B-type atom. This has been previously observed in $TbNi_2Mn$ and $ZrMn_{2+x}$ systems [4, 5]

Pressure-composition isotherms



- Two pressure plateaus are observed for absorption
- The first plateau pressure remains almost unchanged
- The plateau pressure of the second plateau decreases with Mn content
- The C15 phase still not stable upon hydrogen absorption/desorption
- Samples with Mn content between 0.4 and 0.5 show reversible hydrogen absorption due to the presence of an AB_3 phase

CONCLUSIONS

- The C15 Laves phase forms in the $Y_{0.9}Ni_{2-y}Mn_y$ system with large sub-stoichiometry of Y
- PND analysis indicates that the sub-stoichiometry is not due to A-site Y-vacancies as for $Y_{0.95}Ni_2$ but to the anti-site occupation of Mn atom.
- C15 phase containing Mn are not stable upon hydrogenation but AB_3 phase shows good reversibility and very small hysteresis.

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